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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/742,204	12/20/2000	Glen Fox	FUJ 00-01013RAM	7135

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EXAMINER

NGUYEN, KHIEM D

ART UNIT PAPER NUMBER

2823

DATE MAILED: 11/06/2002

Please find below and/or attached an Office communication concerning this application or proceeding.

**Office Action Summary**

Application No.

09/742,204

Applicant(s)

FOX ET AL.

Examiner

Khiem D Nguyen

Art Unit

2823

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

**Period for Reply**

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

**Status**

- 1) ☒ Responsive to communication(s) filed on 22 August 2002.
- 2a) ☒ This action is **FINAL**. 2b) ☐ This action is non-final.
- 3) ☐ Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

**Disposition of Claims**

- 4) ☒ Claim(s) 1-17, 19-24 and 27-31 is/are pending in the application.
- 4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.
- 5) ☐ Claim(s) \_\_\_\_\_ is/are allowed.
- 6) ☒ Claim(s) 1-17, 19-24 and 27-31 is/are rejected.
- 7) ☐ Claim(s) \_\_\_\_\_ is/are objected to.
- 8) ☐ Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

**Application Papers**

- 9) ☐ The specification is objected to by the Examiner.
- 10) ☐ The drawing(s) filed on \_\_\_\_\_ is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
- 11) ☐ The proposed drawing correction filed on \_\_\_\_\_ is: a) ☐ approved b) ☐ disapproved by the Examiner.  
If approved, corrected drawings are required in reply to this Office action.
- 12) ☐ The oath or declaration is objected to by the Examiner.

**Priority under 35 U.S.C. §§ 119 and 120**

- 13) ☐ Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  
a) ☐ All b) ☐ Some \* c) ☐ None of:  
1. ☐ Certified copies of the priority documents have been received.  
2. ☐ Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.  
3. ☐ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).  
\* See the attached detailed Office action for a list of the certified copies not received.
- 14) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).  
a) ☐ The translation of the foreign language provisional application has been received.
- 15) ☐ Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

**Attachment(s)**

- 1) ☒ Notice of References Cited (PTO-892) 4) ☐ Interview Summary (PTO-413) Paper No(s). \_\_\_\_\_
- 2) ☐ Notice of Draftsperson's Patent Drawing Review (PTO-948) 5) ☐ Notice of Informal Patent Application (PTO-152)
- 3) ☐ Information Disclosure Statement(s) (PTO-1449) Paper No(s) \_\_\_\_\_ 6) ☐ Other: \_\_\_\_\_

**DETAILED ACTION**

***New Grounds of Rejection***

***Claim Rejections - 35 USC § 103***

1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.
2. Claims 1 and 4 are rejected under 35 U.S.C. 103(a) as being unpatentable over the applicant's admitted prior art (AAPA) of this application in view of Miyazawa et al. (U.S. Patent 5,953,619) and Evans et al. (U.S. Patent 6,150,184).

AAPA teaches a method for fabrication of ferroelectric capacitor elements of an integrated circuit comprising the steps of (see the Background of the Invention on pages 1-4 of this application):

deposition of an electrically conductive bottom electrode layer;

deposition of a layer of ferroelectric dielectric material, wherein the ferroelectric dielectric layer is comprises PZT;

annealing the layer of ferroelectric dielectric material with a first anneal;

annealing the layer of ferroelectric dielectric material with a second anneal, the second anneal being performed by rapid thermal annealing; and

deposition of an electrically conductive top electrode layer.

AAPA teaches doing the rapid thermal annealing before the formation of the top electrode but fails to teach doing the rapid thermal annealing after the formation of the top electrode as recited in present claims 1.

Miyazawa teaches doing the rapid thermal annealing before or after the upper electrode layer is deposited (See col. 4, line 66 to col. 5, line 8). *It would have been obvious to one of ordinary skill in the art* to incorporate Miyazawa's teaching into AAPA's method because in doing so the PZT dielectric film can be polycrystallized (See col. 5, lines 9-10).

AAPA fails to teach etching the electrically conductive top electrode layer and annealing the layer of ferroelectric dielectric material with another anneal after etching the electrically conductive top electrode layer as recited in present claim 1.

Evans teaches etching the top electrode layer and annealing the layer of ferroelectric layer after etching the top electrode layer (See col. 7, lines 48-56). *It would have been obvious to one of ordinary skill in the art* to incorporate Evans's teaching into AAPA's method because in doing so the electrical switching performance of the ferroelectric capacitors can be improved (See col. 2, lines 1-7).

3. Claims 2-3 and 5-11 are rejected under 35 U.S.C. 103(a) as being unpatentable over the applicant's admitted prior art (AAPA) of this application in view of Miyazawa et al. (U.S. Patent 5,953,619) and Evans et al. (U.S. Patent 6,150,184) as applied to claims 1 and 4 above, and further in view of Joshi et al. (U.S. Patent 6,322,849) and Van Buskirk et al. (U.S. Patent 6,316,797).

AAPA fails to teach that the electrically conductive bottom electrode layer comprises a noble metal and wherein the electrically conductive bottom electrode layer comprises platinum as recited in claims 2 and 3.

However, Joshi teaches that the electrically conductive bottom electrode layer comprises platinum and palladium. See col. 5, lines 11-15. *It would have been obvious to one of ordinary skill in the art* to use platinum as bottom electrode material in the method of AAPA because platinum can provide electrode or contact function.

AAPA fails to teach that the electrically conductive top electrode layer comprises a noble metal oxide and wherein the electrically conductive top electrode layer comprises iridium oxide as recited in present claims 5 and 6.

However, Van Buskirk teaches that top electrode comprises of iridium oxide. See col. 21, lines 5-10. *It would have been obvious to one of ordinary skill in the art* to use iridium oxide as top electrode material in the method of AAPA because iridium oxide can provide electrode or contact function.

AAPA teaches the first and second annealing is done at a temperature and for time duration but fails to teach the ranges for the annealing temperature and time duration as recited in present claims 7-11.

However, it would have been obvious to *one of ordinary skill in the art of making semiconductor devices* to determine the workable or optimal ranges for the annealing temperature and time duration through routine experimentation and optimization to obtain optimal or desired device performance because the annealing temperature and time duration are result-effective variables and there is no evidence

indicating that the annealing temperature and time duration are critical and it has been held that it is not inventive to discover the optimum or workable ranges of a result-effective variable within given prior art conditions by routine experimentation. See MPEP 2144.05.

4. Claims 12-17 and 19-24 are rejected under 35 U.S.C. 103(a) as being unpatentable over the applicant's admitted prior art (AAPA) of this application in view of Miyazawa et al. (U.S. Patent 5,953,619) and Otto et al. (U.S. Patent 6,284,712).

AAPA teaches a method for fabrication of ferroelectric capacitor elements of an integrated circuit comprising the steps of (see the Background of the Invention on pages 1-4 of this application):

deposition of an electrically conductive bottom electrode layer comprising a noble metal;

deposition of a layer of ferroelectric dielectric material, wherein the ferroelectric dielectric layer is comprises PZT and performed by sputtering;

annealing the layer of ferroelectric dielectric material with a first anneal;

deposition of an electrically conductive top electrode layer comprising a noble metal oxide;

depositing an encapsulation layer; and,

annealing the layer of ferroelectric dielectric material with a second anneal, the second anneal being performed by rapid thermal annealing and performed after the step of deposition of an electrically conductive top electrode layer; and,

AAPA teaches doing the rapid thermal annealing before the formation of the top electrode but fails to teach doing the rapid thermal annealing after the formation of the top electrode as recited in present claim 12.

Miyazawa teaches doing the rapid thermal annealing before or after the upper electrode layer is deposited (See col. 4, line 66 to col. 5, line 8). *It would have been obvious to one of ordinary skill in the art* to incorporate Miyazawa's teaching into AAPA's method because in doing so the PZT dielectric film can be polycrystallized (See col. 5, lines 9-10).

AAPA fails to teach that the first and second anneal is performed in an environment comprising a mixture of oxygen and inert gas as recited in present claims 12 and 15-20.

Otto teaches that the annealing process is performed in an environment comprising a mixture of inert gas and oxygen (See col. 14, lines 59-64). *It would have been obvious to one of ordinary skill in the art* to performed the first anneal in an environment comprising a mixture of oxygen and inert gas because doing so the desired total oxygen pressure can be obtained (See col. 14, lines 59-65).

AAPA fails to teach the ranges for the partial pressure of oxygen and the annealing temperature as recited in claims 12, 15-16, 19 and 22.

However, it would have been obvious to *one of ordinary skill in the art of making semiconductor devices* to determine the workable or optimal ranges for the partial pressure of oxygen and the annealing temperature through routine experimentation and optimization to obtain optimal or desired device performance because the partial

pressure of oxygen and the annealing temperature are result-effective variables and there is no evidence indicating that the partial pressure of oxygen and the annealing temperature are critical and it has been held that it is not inventive to discover the optimum or workable ranges of a result-effective variable within given prior art conditions by routine experimentation. See MPEP 2144.05.

5. Claims 27-31 are rejected under 35 U.S.C. 103(a) as being unpatentable over the applicant's admitted prior art (AAPA) of this application in view of Miyazawa et al. (U.S. Patent 5,953,619) and Van Buskirk et al. (U.S. Patent 6,316,797).

AAPA teaches a method for fabrication of ferroelectric capacitor elements of an integrated circuit comprising the steps of (see the Background of the Invention on pages 1-4 of this application):

deposition of an electrically conductive bottom electrode layer comprising a noble metal;

deposition of a layer of ferroelectric dielectric material, wherein the ferroelectric dielectric layer is comprises PZT and performed by sputtering;

annealing the layer of ferroelectric dielectric material with a first anneal;

deposition of an electrically conductive top electrode layer comprising a noble metal oxide; and

annealing the layer of ferroelectric dielectric material with a second anneal, the second anneal being performed by rapid thermal annealing and performed after the step of deposition of an electrically conductive top electrode layer.



AAPA teaches doing the rapid thermal annealing before the formation of the top electrode but fails to teach doing the rapid thermal annealing after the formation of the top electrode wherein the anneal changing the layer of ferroelectric material into grains having a columnar structure as recited in present claims 27 and 30.

Miyazawa teaches doing the rapid thermal annealing before or after the upper electrode layer 29a is deposited wherein the anneal changing the layer of ferroelectric material 28 into grains having a columnar structure (See col. 4, line 66 to col. 5, line 8 and FIGS 10-12). *It would have been obvious to one of ordinary skill in the art* to incorporate Miyazawa's teaching into AAPA's method because in doing so the PZT dielectric film can be polycrystallized (See col. 5, lines 9-10).

AAPA fails to teach that the electrically conductive top electrode layer comprises a noble metal oxide and wherein the electrically conductive top electrode layer comprises iridium oxide as recited in present claims 29 and 30.

However, Van Buskirk teaches that top electrode comprises of iridium oxide (See col. 21, lines 5-10). *It would have been obvious to one of ordinary skill in the art* to use iridium oxide as top electrode material in the method of AAPA because iridium oxide can provide electrode or contact function.

### ***Response to Amendment***

### ***Responding to applicant's Arguments***

6. Applicant's arguments with respect to claims 1-17 and 19-24 have been considered but are moot in view of the new ground(s) of rejection.

### ***Conclusion***

7. Applicant's amendment necessitated the new ground(s) of rejection presented in this Office action. Accordingly, **THIS ACTION IS MADE FINAL**. See MPEP § 706.07(a). Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire **THREE MONTHS** from the mailing date of this action. In the event a first reply is filed within **TWO MONTHS** of the mailing date of this final action and the advisory action is not mailed until after the end of the **THREE-MONTH** shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than **SIX MONTHS** from the date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Khiem D Nguyen whose telephone number is (703) 306-0210. The examiner can normally be reached on Monday-Friday (8:00 AM - 5:00 PM).


If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Chaudhuri Olik can be reached on (703) 306-2794. The fax phone numbers for the organization where this application or proceeding is assigned are (703) 746-9179 for regular communications and (703) 746-9179 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is (703) 308-0956.

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K.N.  
November 1, 2002



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